**Fractional Distillation (Level 1) examiners tips: Read these please!**

**•** larger/bigger/heavier molecules condense at the bottom of the fractionating column…

where there are higher temperatures…

because larger/bigger/heavier molecules have a greater/stronger attraction towards each other  
**•** you MUST link boiling point to the size of molecules and collection point.

**Also…”don’t be daft”**

hydrocarbons contain....believe it or not...HYDROGEN + CARBON only

boiling point is the change in state from a liquid to a gas

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